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## Key indicators

Single-crystal X-ray study
$T=293 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.008 \AA$
$R$ factor $=0.054$
$w R$ factor $=0.165$
Data-to-parameter ratio $=35.8$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## Chloro(methyl)( $\boldsymbol{\eta}^{5}$-pentamethylcyclopentadienyl)(trimethylphosphine)rhodium(III)

The title compound, $\mathrm{RhCp} * \mathrm{Me}(\mathrm{Cl})\left(\mathrm{PMe}_{3}\right)$ or $\left[\mathrm{RhCl}\left(\mathrm{CH}_{3}\right)\right.$ $\left(\mathrm{C}_{10} \mathrm{H}_{15}\right)\left(\mathrm{C}_{3} \mathrm{H}_{9} \mathrm{P}\right)$, where $\mathrm{Cp} *$ is pentamethylcyclopentadienyl, has a pseudo-tetrahedral coordination geometry. The Cp* ligand is planar and $\eta^{5}$-coordinated. The molecular structure can be described as a so-called three-legged pianostool. The rhodium-methyl, rhodium-chlorine and rhodiumphosphorus distances are $2.255(4), 2.3764$ (18) and 2.2436 (13) Å, respectively.

## Comment

The title compound, (I) (Fig. 1), was first prepared by Jones \& Feher (1984) and belongs to a group of complexes that have been used extensively in the activation of small molecules (Lefort et al., 1998; Arndtsen et al., 1995). However, there are no reported crystal structures of chloro-methyl complexes of this type.

(I)

The coordination geometry around rhodium is pseudotetrahedral and the $\mathrm{Cp} *$ ligand is planar and $\eta^{5}$-coordinated. This gives rise to a three-legged piano-stool where the angles around rhodium involving the monodentate ligands are


Figure 1
Numbering scheme with displacement ellipsoids ( $30 \%$ probability) for the title compound.

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around $88^{\circ}$. The deviation from idealized tetrahedral angles is probably dictated by the bulky Cp* ligand. Selected bond lengths and angles are given in Table 1. The closest contact between complexes is 2.04 (1) $\AA$ and is found between $\mathrm{H} 9 A$ and $\mathrm{H} 9 A(-x,-y, 1-z)$. This short distance is probably an artefact that arises from the fact that H atoms are placed only to minimize intramolecular interactions. The $\mathrm{C} 9 \ldots \mathrm{C} 9(-x,-y$, $1-z$ ) distance is 3.870 (6) $\AA$ and rotation around the $\mathrm{Rh}-\mathrm{C} 9$ bond makes the $\mathrm{C}-\mathrm{H}$ distances much longer.

## Experimental

$\mathrm{RhCp} * \mathrm{Me}_{2}\left(\mathrm{PMe}_{3}\right)$ was synthesized according to Jones \& Feher (1984) and recrystallization from petroleum ether afforded red crystals of the title compound as a by-product.

## Crystal data

| $\left[\mathrm{RhCl}_{2}\left(\mathrm{CH}_{3}\right)\left(\mathrm{C}_{10} \mathrm{H}_{15}\right)\left(\mathrm{C}_{3} \mathrm{H}_{9} \mathrm{P}\right)\right]$ | Mo $K \alpha$ radiation |
| :--- | :--- |
| $M_{r}=364.69$ | Cell parameters from 6294 |
| Orthorhombic, $P b c a$ | reflections |
| $a=8.9295(18) \AA$ | $\theta=2.8-25.4^{\circ}$ |
| $b=14.792(3) \AA$ | $\mu=1.23 \mathrm{~mm}^{-1}$ |
| $c=25.812(5) \AA$ | $T=293(2) \mathrm{K}$ |
| $V=3409.2(12) \AA^{3}$ | Prism, red |
| $Z=8$ | $0.15 \times 0.03 \times 0.03 \mathrm{~mm}$ |
| $D_{x}=1.421 \mathrm{Mg} \mathrm{m}^{-3}$ |  |

## Data collection

Bruker SMART CCD
diffractometer
$\omega$ scans
Absorption correction: empirical
(SADABS; Sheldrick, 1996)
$T_{\text {min }}=0.789, T_{\text {max }}=0.943$
33212 measured reflections
5506 independent reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.054$
$w R\left(F^{2}\right)=0.165$
$S=1.00$
5506 reflections
154 parameters

Table 1
Selected geometric parameters $\left(\AA^{\circ},{ }^{\circ}\right)$.

| Rh1-C4 | $2.179(4)$ | $\mathrm{P} 1-\mathrm{C} 8$ | $1.807(6)$ |
| :--- | :--- | :--- | :--- |
| Rh1-C5 | $2.225(4)$ | $\mathrm{P} 1-\mathrm{C} 7$ | $1.822(6)$ |
| Rh1-C3 | $2.225(5)$ | $\mathrm{P} 1-\mathrm{C} 6$ | $1.823(5)$ |
| Rh1-C2 | $2.228(5)$ | $\mathrm{C} 1-\mathrm{C} 5$ | $1.415(6)$ |
| Rh1-C1 | $2.230(4)$ | $\mathrm{C} 1-\mathrm{C} 2$ | $1.453(8)$ |
| Rh1-P1 | $2.2436(13)$ | $\mathrm{C} 2-\mathrm{C} 3$ | $1.344(8)$ |
| Rh1-C9 | $2.255(4)$ | $\mathrm{C} 3-\mathrm{C} 4$ | $1.419(9)$ |
| Rh1-Cl1 | $2.3764(18)$ | $\mathrm{C} 4-\mathrm{C} 5$ | $1.393(7)$ |
|  |  |  |  |
| P1-Rh1-C9 | $89.02(11)$ | $\mathrm{C} 7-\mathrm{P} 1-\mathrm{Rh} 1$ | $116.3(2)$ |
| P1-Rh1-Cl1 | $87.88(6)$ | $\mathrm{C} 6-\mathrm{P} 1-\mathrm{Rh} 1$ | $116.97(19)$ |
| C9-Rh1-Cl1 | $87.55(12)$ | $\mathrm{C} 5-\mathrm{C} 1-\mathrm{C} 2$ | $105.1(4)$ |
| $\mathrm{C} 8-\mathrm{P} 1-\mathrm{C} 7$ | $103.3(3)$ | $\mathrm{C} 3-\mathrm{C} 2-\mathrm{C} 1$ | $109.3(5)$ |
| C8-P1-C6 | $103.2(3)$ | $\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4$ | $109.0(5)$ |
| C7-P1-C6 | $102.7(3)$ | $\mathrm{C} 5-\mathrm{C} 4-\mathrm{C} 3$ | $107.4(5)$ |
| C8-P1-Rh1 | $112.6(2)$ | $\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 1$ | $109.0(4)$ |

The highest residual electron density is located within $0.5 \AA$ of C 9 and Cl1.

Data collection: SMART (Bruker, 1995); cell refinement: SAINT (Bruker, 1995); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: DIAMOND (Brandenburg, 1999); software used to prepare material for publication: SHELXL97.

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