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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.008 Å R factor = 0.054 wR factor = 0.165 Data-to-parameter ratio = 35.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Chloro(methyl)(η^5 -pentamethylcyclopentadienyl)-(trimethylphosphine)rhodium(III)

The title compound, RhCp*Me(Cl)(PMe₃) or [RhCl(CH₃)-(C₁₀H₁₅)(C₃H₉P)], where Cp* is pentamethylcyclopentadienyl, has a pseudo-tetrahedral coordination geometry. The Cp* ligand is planar and η^5 -coordinated. The molecular structure can be described as a so-called three-legged pianostool. The rhodium–methyl, rhodium–chlorine and rhodium– phosphorus distances are 2.255 (4), 2.3764 (18) and 2.2436 (13) Å, respectively. Received 3 October 2001 Accepted 9 October 2001 Online 13 October 2001

Comment

The title compound, (I) (Fig. 1), was first prepared by Jones & Feher (1984) and belongs to a group of complexes that have been used extensively in the activation of small molecules (Lefort *et al.*, 1998; Arndtsen *et al.*, 1995). However, there are no reported crystal structures of chloro–methyl complexes of this type.



The coordination geometry around rhodium is pseudotetrahedral and the Cp* ligand is planar and η^5 -coordinated. This gives rise to a three-legged piano-stool where the angles around rhodium involving the monodentate ligands are



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Figure 1

Numbering scheme with displacement ellipsoids (30% probability) for the title compound.

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around 88°. The deviation from idealized tetrahedral angles is probably dictated by the bulky Cp* ligand. Selected bond lengths and angles are given in Table 1. The closest contact between complexes is 2.04 (1) Å and is found between H9A and H9A(-x, -y, 1-z). This short distance is probably an artefact that arises from the fact that H atoms are placed only to minimize intramolecular interactions. The C9 \cdots C9(-x, -y, 1-z) distance is 3.870 (6) Å and rotation around the Rh–C9 bond makes the C–H distances much longer.

Experimental

 $RhCp*Me_2(PMe_3)$ was synthesized according to Jones & Feher (1984) and recrystallization from petroleum ether afforded red crystals of the title compound as a by-product.

Crystal data

 $[RhCl(CH_3)(C_{10}H_{15})(C_3H_9P)]$ $M_r = 364.69$ Orthorhombic, *Pbca* a = 8.9295 (18) Å b = 14.792 (3) Å c = 25.812 (5) Å V = 3409.2 (12) Å³ Z = 8 $D_x = 1.421$ Mg m⁻³

Data collection

Bruker SMART CCD diffractometer ω scans Absorption correction: empirical (*SADABS*; Sheldrick, 1996) $T_{min} = 0.789, T_{max} = 0.943$ 33 212 measured reflections 5506 independent reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.054$ $wR(F^2) = 0.165$ S = 1.005506 reflections 154 parameters Mo $K\alpha$ radiation Cell parameters from 6294 reflections $\theta = 2.8-25.4^{\circ}$ $\mu = 1.23 \text{ mm}^{-1}$ T = 293 (2) K Prism, red 0.15 × 0.03 × 0.03 mm

2862 reflections with $I > 2\sigma(I)$ $R_{int} = 0.069$ $\theta_{max} = 31.9^{\circ}$ $h = -12 \rightarrow 13$ $k = -20 \rightarrow 21$ $l = -36 \rightarrow 37$ Intensity decay: none

H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.0873P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{\rm max} = 0.001$
$\Delta \rho_{\rm max} = 0.94 \text{ e} \text{ Å}^{-3}$
$\Delta \rho_{min} = -1.36 \text{ e} \text{ Å}^{-3}$

Table 1

Selected geometric parameters (Å, °).

Rh1-C4	2.179 (4)	P1-C8	1.807 (6)
Rh1-C5	2.225 (4)	P1-C7	1.822 (6)
Rh1-C3	2.225 (5)	P1-C6	1.823 (5)
Rh1-C2	2.228 (5)	C1-C5	1.415 (6)
Rh1-C1	2.230 (4)	C1-C2	1.453 (8)
Rh1-P1	2.2436 (13)	C2-C3	1.344 (8)
Rh1-C9	2.255 (4)	C3-C4	1.419 (9)
Rh1-Cl1	2.3764 (18)	C4-C5	1.393 (7)
P1-Rh1-C9	89.02 (11)	C7-P1-Rh1	116.3 (2)
P1-Rh1-Cl1	87.88 (6)	C6-P1-Rh1	116.97 (19)
C9-Rh1-Cl1	87.55 (12)	C5-C1-C2	105.1 (4)
C8-P1-C7	103.3 (3)	C3-C2-C1	109.3 (5)
C8-P1-C6	103.2 (3)	C2-C3-C4	109.0 (5)
C7-P1-C6	102.7 (3)	C5-C4-C3	107.4 (5)
C8-P1-Rh1	112.6 (2)	C4-C5-C1	109.0 (4)

The highest residual electron density is located within 0.5 Å of C9 and Cl1.

Data collection: *SMART* (Bruker, 1995); cell refinement: *SAINT* (Bruker, 1995); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *DIAMOND* (Brandenburg, 1999); software used to prepare material for publication: *SHELXL*97.

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